



Graphene sensors – detection of gases

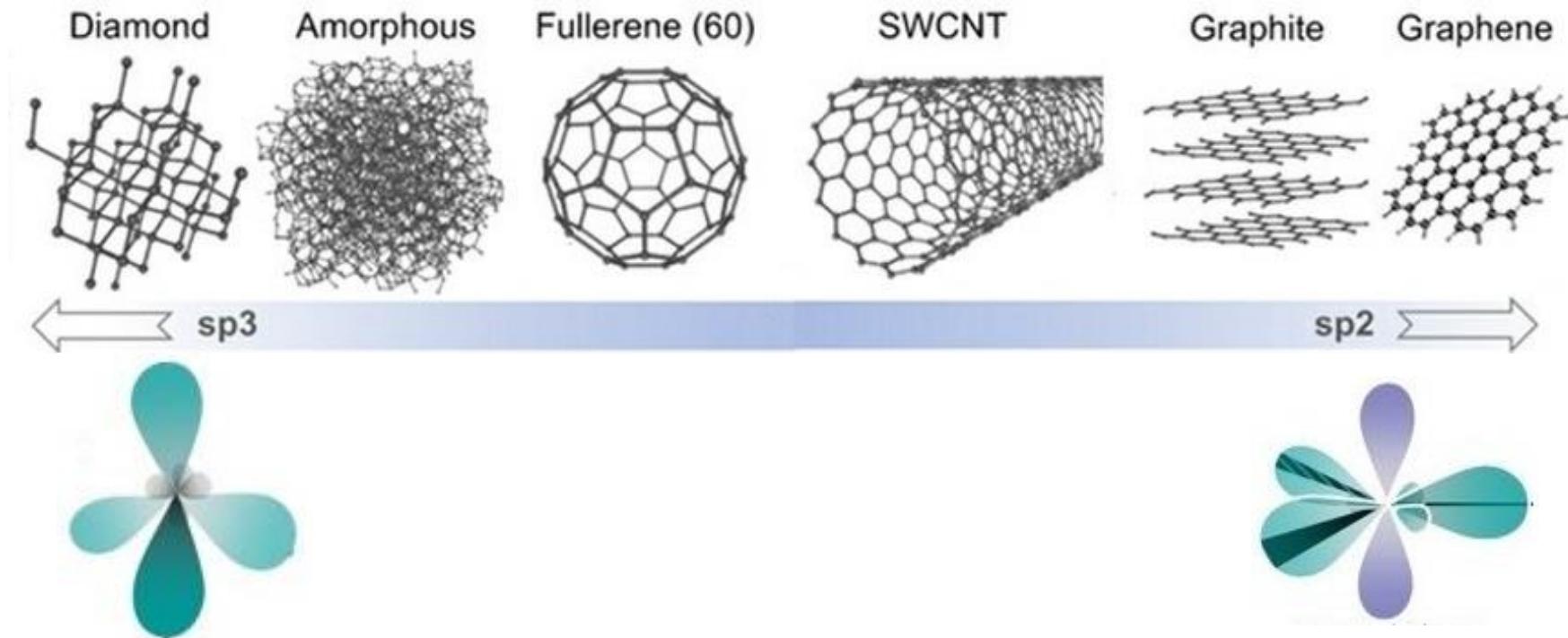
Chapters of Nanostructures
TU Liberec

Author: Jaroslav Grof
Supervisor: Ing. Jiří Červenka, Ph.D.
26. 5. 2020

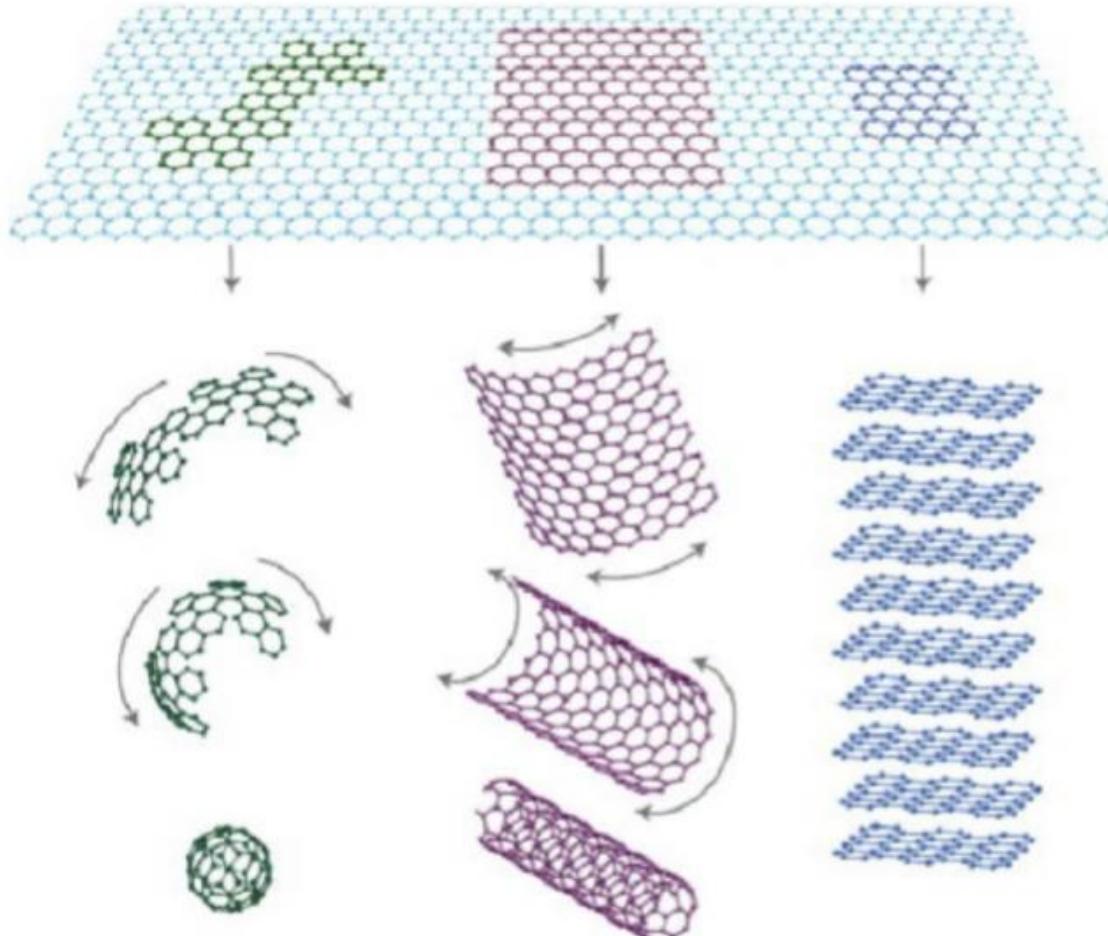




Allotropes of carbon



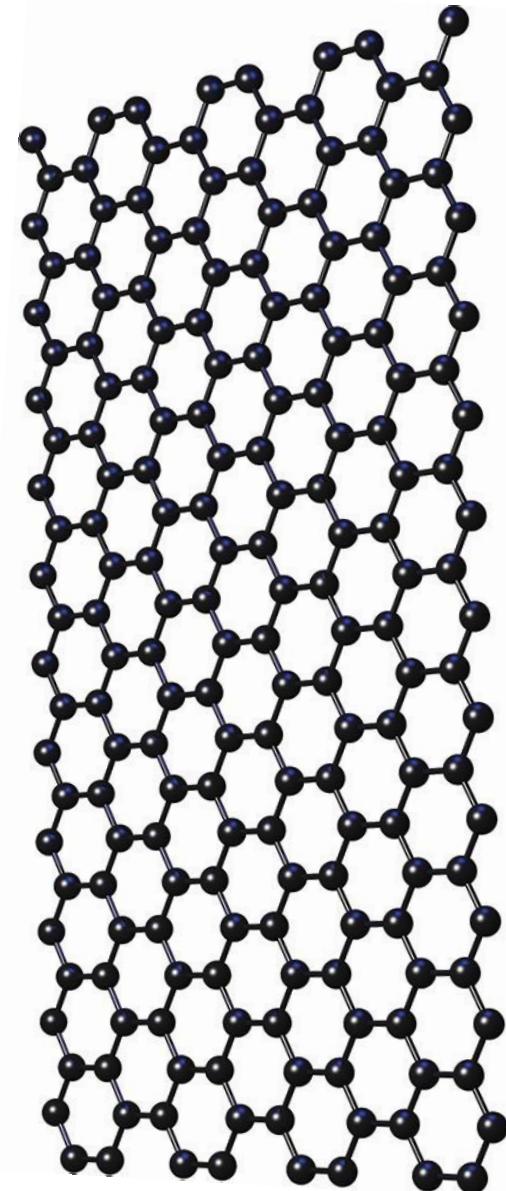
Allotropes of carbon



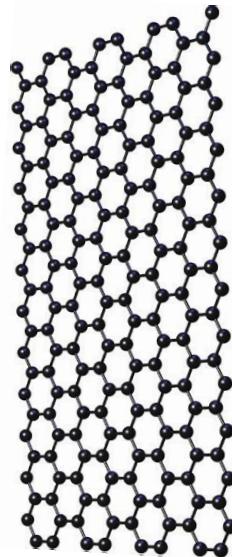
NAG, Anindya, 2018. *Sensors and Actuators A: Physical*.

Graphene

- Allotrope of carbon
- Single layer of atoms
- Two-dimensional hexagonal lattice
- Indefinitely large aromatic molecule
- Ultimate case of polycyclic aromatic hydrocarbons
- sp^2



ZHAN, Beibei, 2014. *Small.*



Unique properties of graphene

2 630 m²·g⁻¹

0,76 mg·m⁻²

1,1 TPa

5 000 W·m⁻¹·K⁻¹

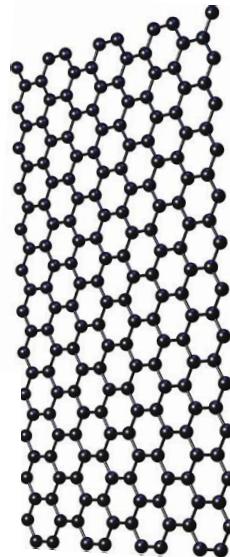
2,3% absorption

10⁸ S·m⁻¹

4 510 K

1,6·10⁹ A·cm⁻²

nanosilica 1 200 m²·g⁻¹



Unique properties of graphene

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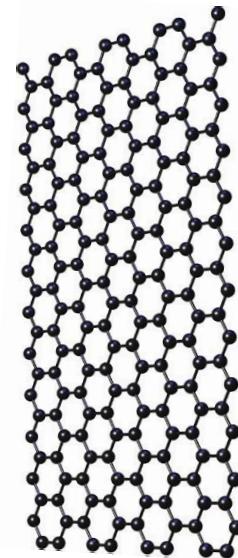
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$10^8\ S \cdot m^{-1}$

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$1,6 \cdot 10^9\ A \cdot cm^{-2}$





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steel 0,2 TPa

diamond 1 000 – 2 300

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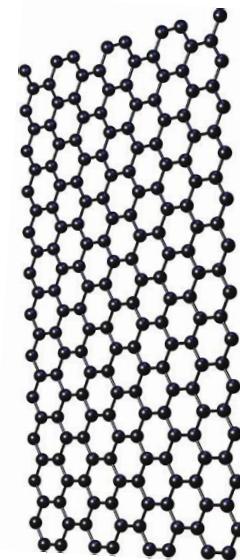
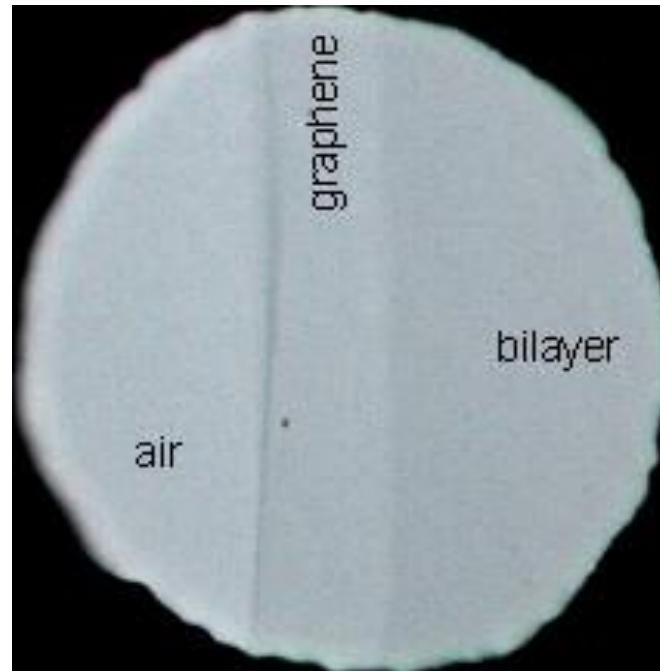
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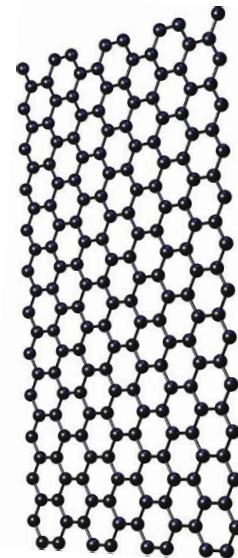
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steel





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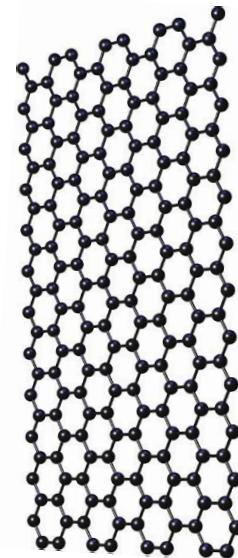
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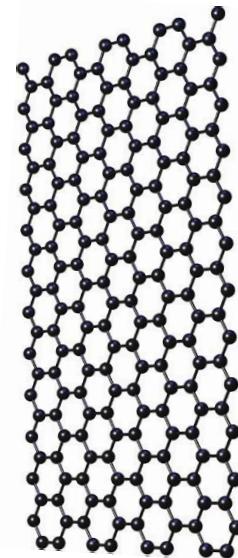
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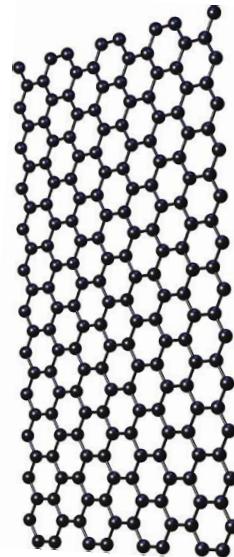
steel 0,2 TPa

diamond 1 000 – 2 300

silver 6,3·10⁷ S·m⁻¹

tungsten 3 695 K

copper 10⁵ A·cm⁻²

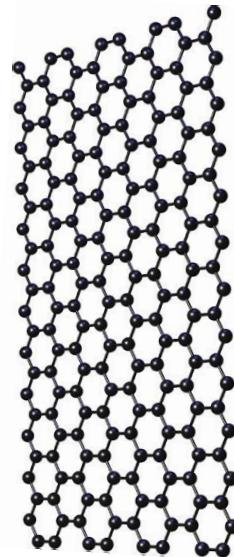


Preparation of graphene

Hummer's method

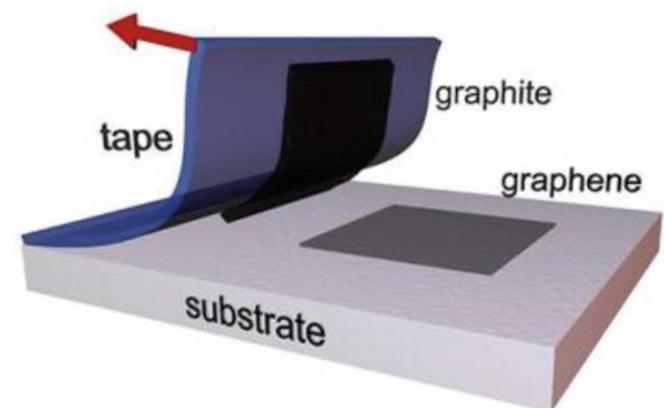
graphite powder washed in aqua regia, heating, water treatment, drying > $9\text{ H}_2\text{SO}_4 + 1\text{ H}_3\text{PO}_4$ > slow addition of KMnO_4 , stirring 6 h > H_2O_2 , cooling > washed with HCl + DI water three times > intense heat > sonification > heating > $\text{NaHSO}_3 + \text{Na}_2\text{9H}_2\text{O} + \text{SO}_2$, stirring > filtration, DI water > freeze drying

Preparation of graphene

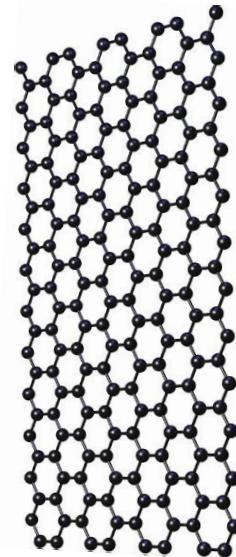


Mechanical exfoliation

- adhesive tape stucked over graphite crystals or flakes leads to entrapment of graphite layers
- multiple steps, each produces fewer layer
- deposition on silicon wafer
- crystallites larger than 1 mm, visible



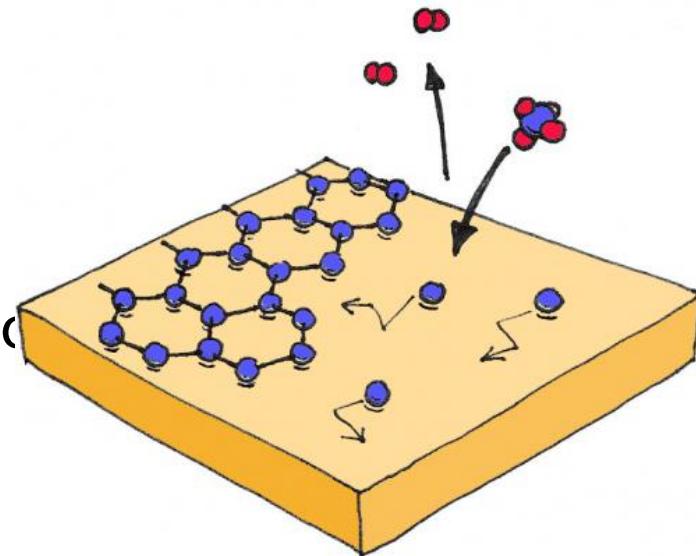
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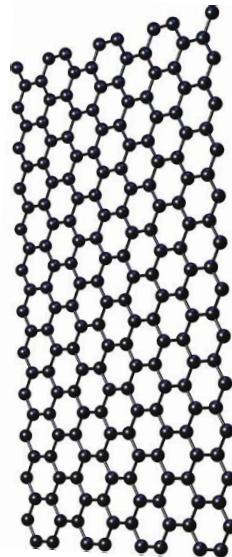


Preparation of graphene

CVD

- quartz furnace with inert environment is heated
- carbonaceous gases Ar/H₂/CH₄
- deposition of carbon on the metal (Ni/Cu/Co/Pt/Ir)
- formation of single long atom-thick monolayer of graphene

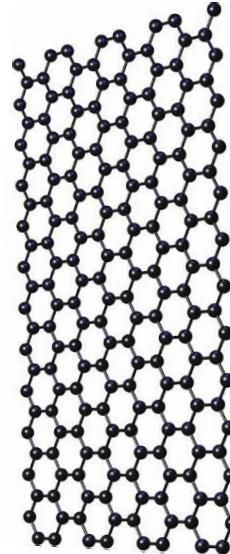




Application of graphene

EVERYWHERE

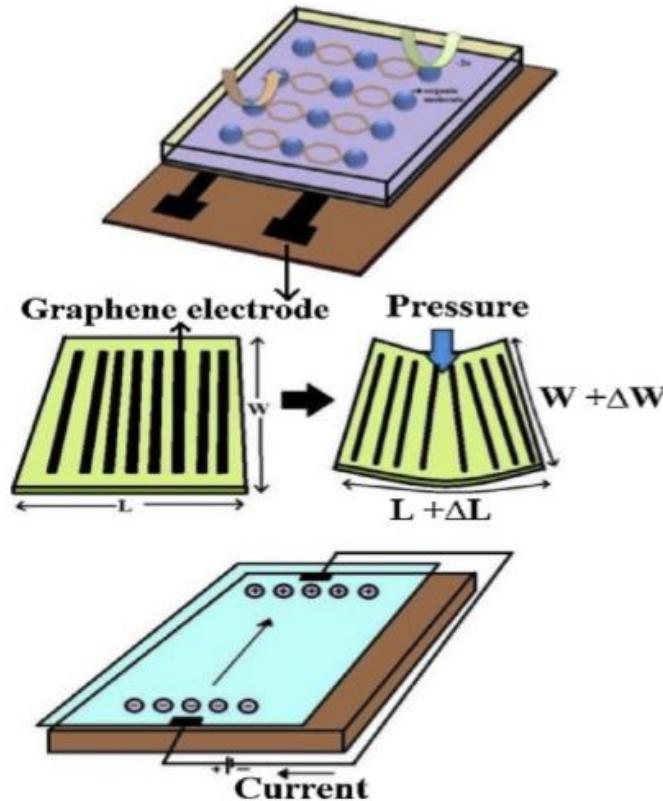
- **Medicine** (contrast agent, drug delivery, ...)
- **Electronics** (transistors, quantum dots, conductive inks, organic electronics, transparent electrodes, ...)
- **Energy** (solar cells, storage)
- **Environmental** (pollution removal, water filtration, ...)
- **Composites**



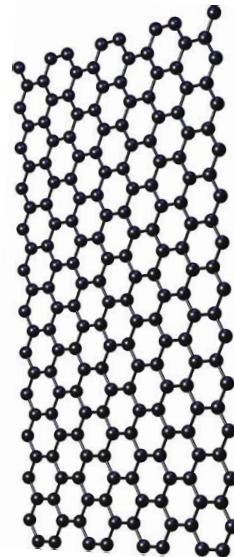
Graphene-based sensors

Types of sensors

- Electrochemical
- Strain
- Electrical



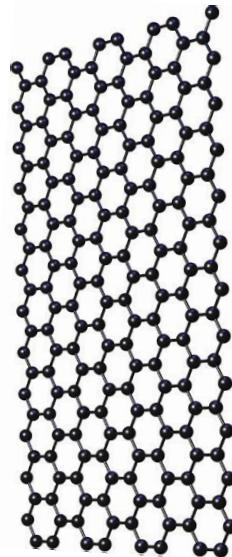
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Graphene-based gas sensors

Gas sensors

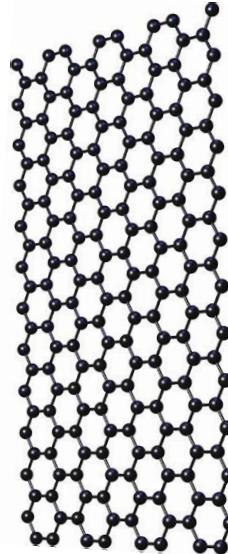
- electronic device that can qualitatively or quantificationally detect specific gases
- conventionally constructed by semiconductor metal oxides
 - drawbacks: high-temperature operation, large power consumtion, low selectivity
 - or conducting polymers
 - degradation



Graphene-based gas sensors

Why?

- all atoms are exposed to surface interactions with gas molecules
- room temperature, strong and flexible
- easy functionalization is able to improve sensing performance

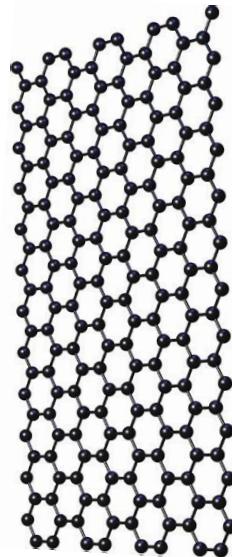


Graphene-based gas sensors

Mechanism

- gaseous adsorbates interact with graphene
- charge transfer processes between gas molecules and graphene surface
 - redistributon of electrons (H_2O)
 - change in electron concentration (NO_2)
 - covalent bonding ($H\cdot$, $HO\cdot$)
- leads to conductance changes

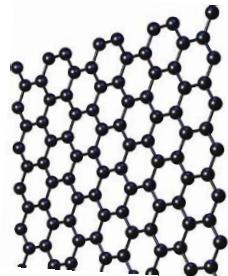
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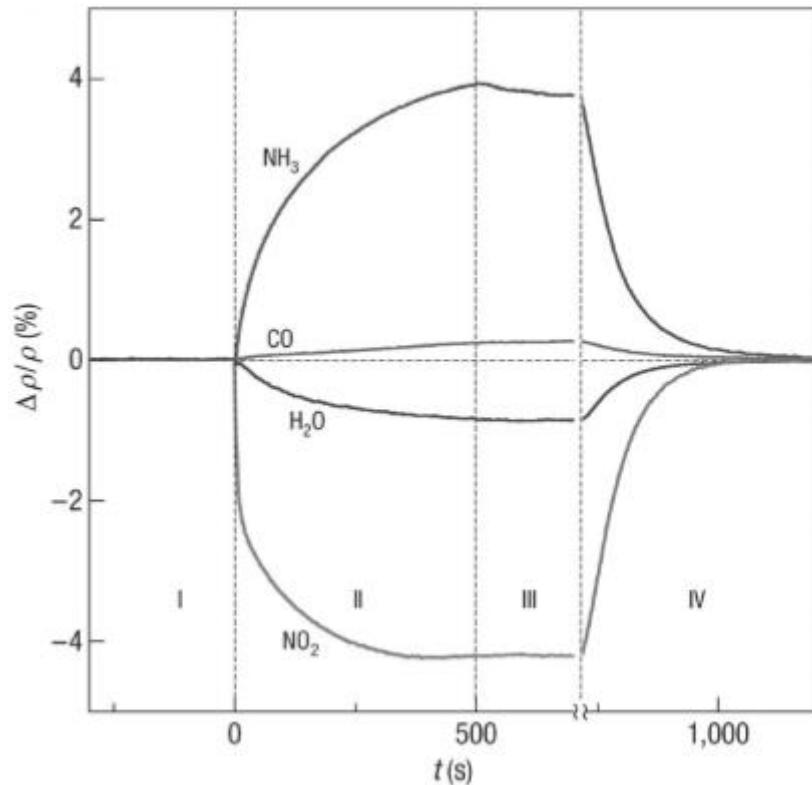
Types

- Chemiresistors – simply, sensitive, stable
 - sensing layer between electrodes
 - resistance/current monitoring
- FET (field effect transistor)
 - source, drain, gate
 - observing output characteristics

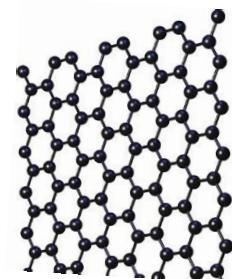


Graphene-based gas sensors

- Changes in resistivity
- (II) 5 L of diluted gas
1 ppm
- (III) evacuation
- (IV) annealing at 150 °C

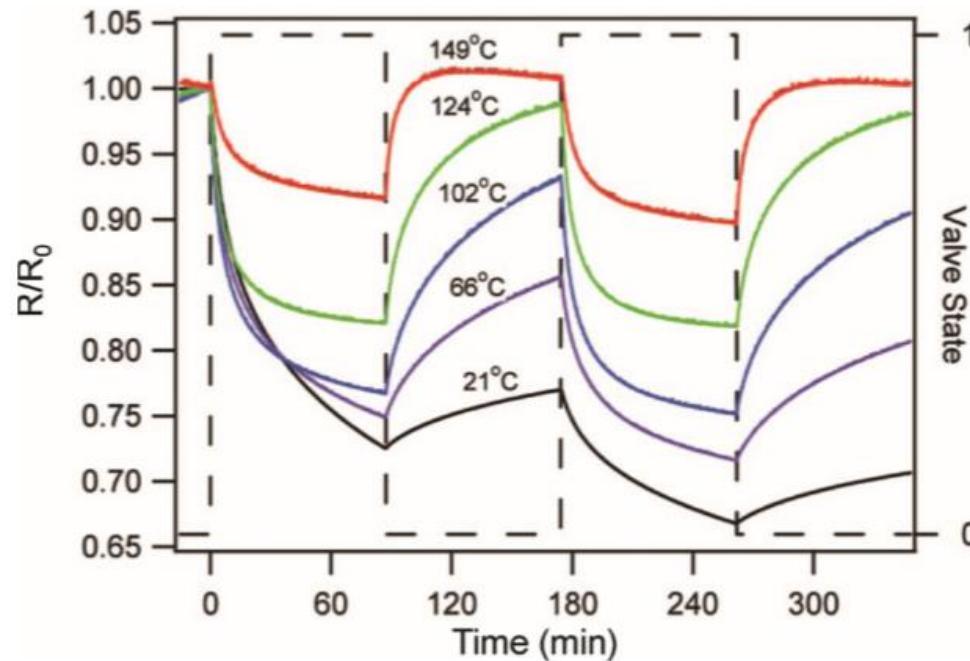


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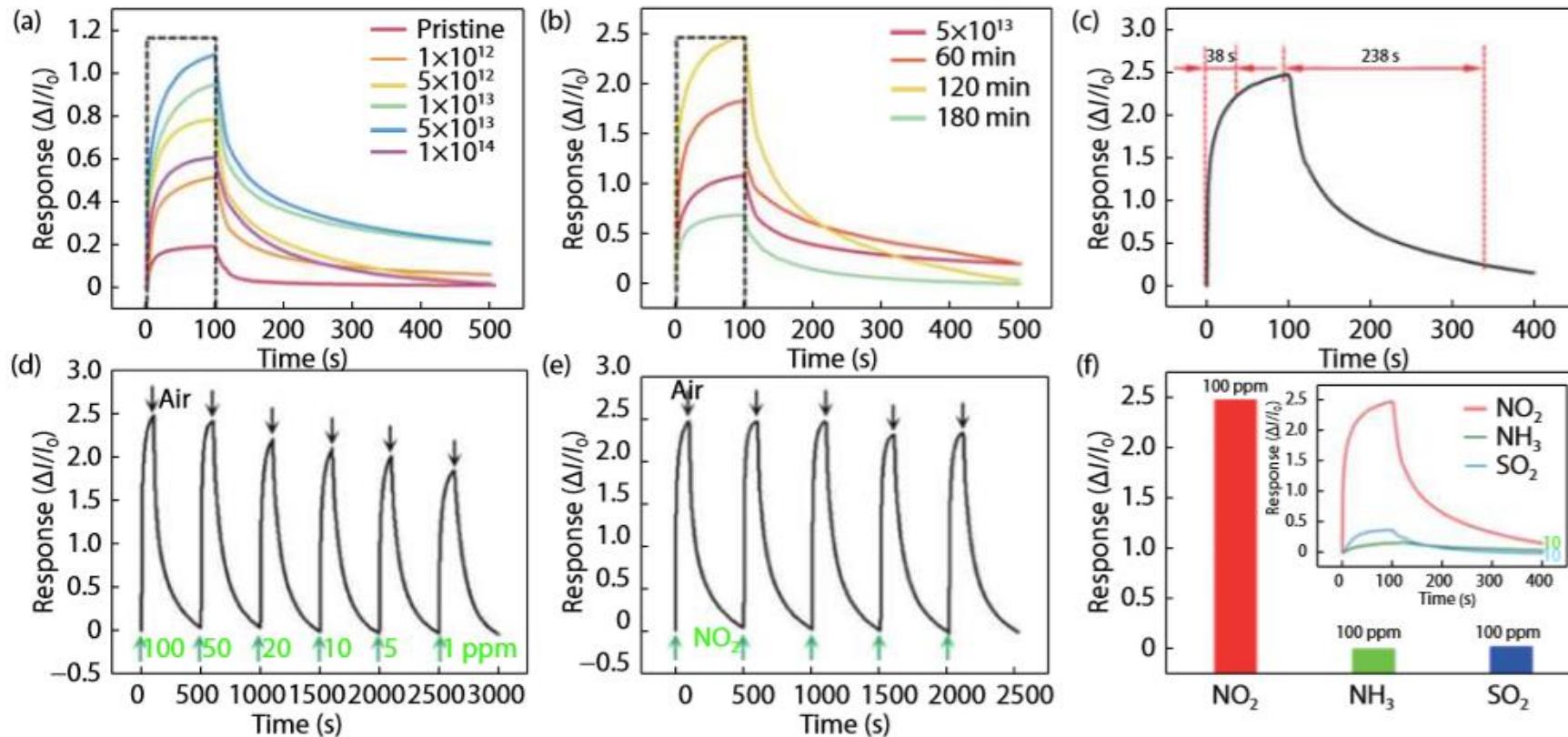
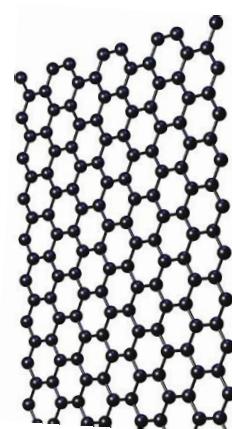
Graphene-based gas sensors

- Changes in resistivity
- Variable temperatures
- 5 ppm of NO₂

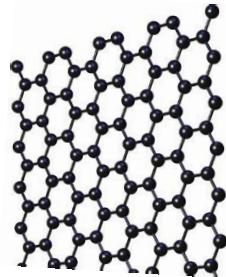


Graphene-based gas sensors

Defective Graphene / Pristine Graphene hybrid layer



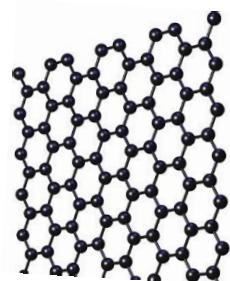
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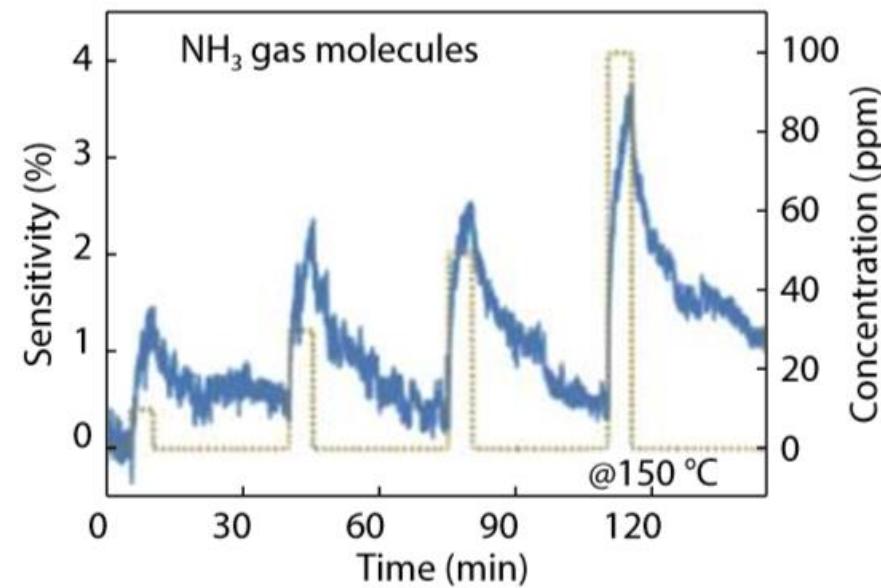
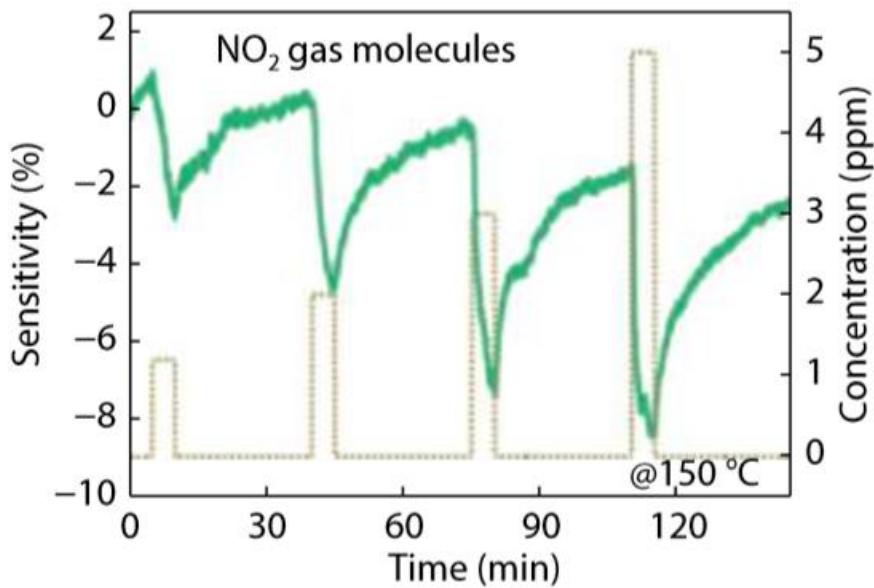
Material	Device type	Synthesis method	Substrate	Analyte	Limit of detection	Working temperature	Response (recovery) time
Graphene + MoS ₂	Resistive	CVD + mechanical exfoliation	Polyimide	NO ₂	1.2 ppm	150 °C	30 min
Graphene + MoS ₂	Resistive	Liquid-phase co-exfoliation	Si/SiO ₂	Methanol	10 ppm	–	210 s (220 s)
Graphene + MoS ₂	Resistive	GA + ATM	Poly-Si	NO ₂	50 ppb	25 °C	21.6 s (< 29.4 s)
Graphene + MoS ₂	FET	CVD + mechanical exfoliation	Si/SiO ₂	NO ₂	1 ppm	RT	–
rGO + MoS ₂	Resistive	Microwave-assisted exfoliation	PDMS	NH ₃	0.48 mbar	RT	15 s
rGO + MoS ₂	Resistive	Soft lithographic patterning	PET	NO ₂	0.15 ppm	90 °C	–
rGO + MoS ₂	Resistive	Lithography	SiO ₂ /Si	NO ₂	2 ppm	60 °C	30 min
rGO + MoS ₂	Resistive	Layer-by-layer self-assembly	SiO ₂ /Si	Formaldehyde	2.5 ppm	RT	73 s
rGO + MoS ₂	Resistive	Self-assembly	PEN	Formaldehyde	2.5 ppm	RT	10 min (13 min)
MoS ₂ /WS ₂	Resistive	Hydrothermal process	–	NO ₂	10 ppb	RT	1.6 s (27.7 s)
rGO/WS ₂	Resistive	Ball milling and sonication	Si ₃ N ₄	NO ₂	1 ppm	RT	22 min (26 min)
Defective graphene/pristine graphene	Current	APCVD	Ge	NO ₂	1 ppm	RT	28 s (238 s)
rGO-MoS ₂ -CdS BP/h-BN/MoS ₂	Resistive FET	Solvothermal Mechanically exfoliated + e-beam lithography	– SiO ₂ /Si	NO ₂ NO ₂	0.2 ppm 3.3 ppb	75 °C RT	25 s (34 s) 8 min (8 min)

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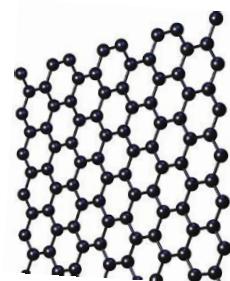


Graphene-based gas sensors

- Graphene + MoS₂
(conductive layer + analyte acceptor)

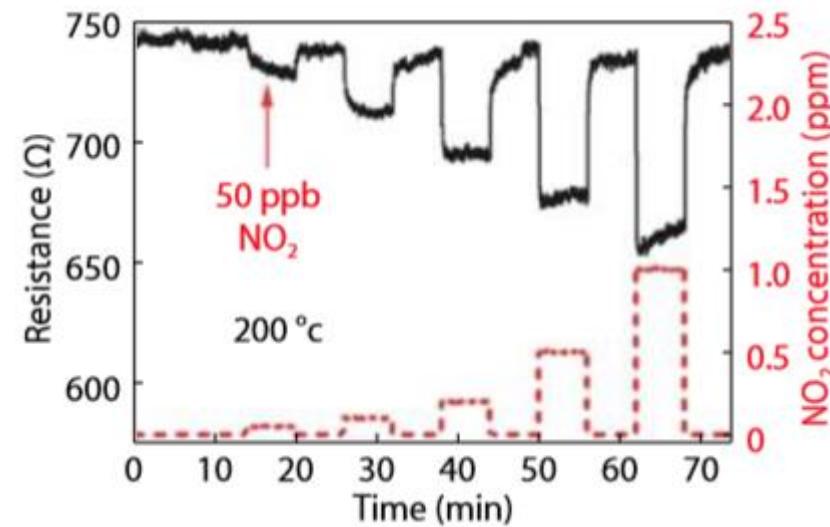
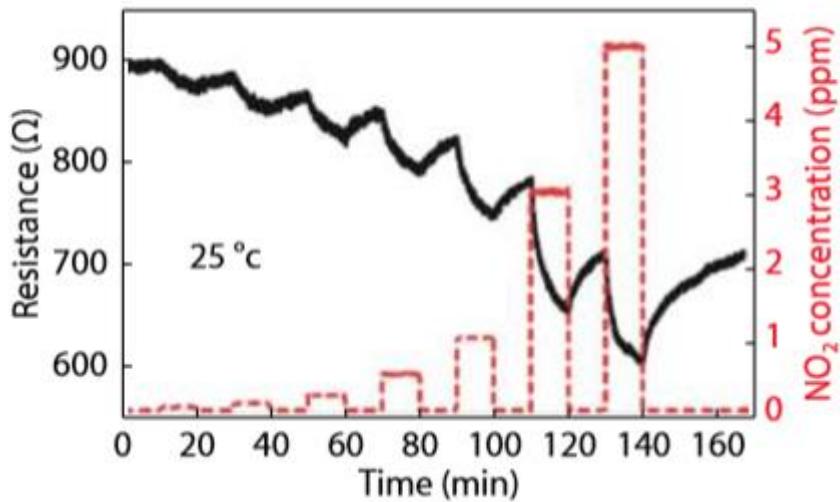


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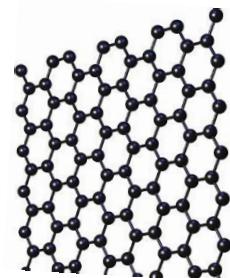


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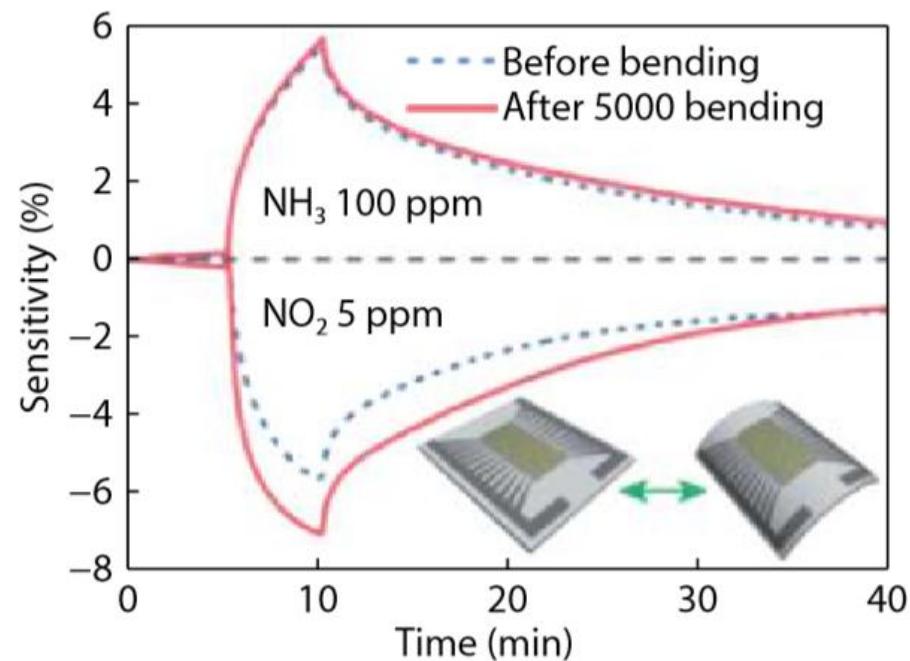


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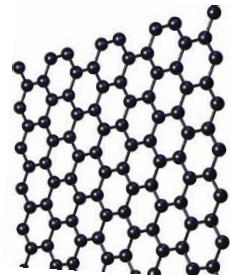


Graphene-based gas sensors

- Graphene + MoS₂
(conductive layer + analyte acceptor)
- Polyimide substrate
- 5 000 bending cycles

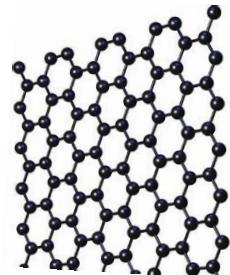


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Summary

- Sensitivity:
1 ppm, special cases 1 ppb
- Selectivity:
 DGr/Gr
65x higher response for NO_2
- Response/
recovery time:
tens/
hundreds of seconds
- Material stability:
temperature, humidity
- Reproducibility



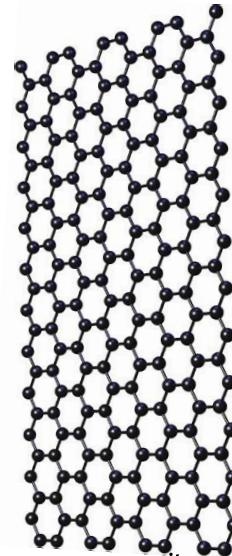
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Thank you for your attention.



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